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488 CHAPTER 16 Solids,
Liquids, and Gases Kinetic
Theory SECTION States of
Matter If you don't finish
lunch quickly, you'll be
late for practice. The soup
is boiling on the stove. You
hastily pour the soup into
the bowl, but now it's too
hot to eat. You add an ice
cube and stir. The soup's
temperature drops—now you
can eat it without burn-

Chapter 16: Solids, Liquids,
and Gases

Physical Science Chapter 16
The Periodic Table Parts of
an Atom • An atom consists
of a nucleus surrounded by

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Answers Navigation

- one or more electrons
- Atoms are electrically neutral w/ the same number of protons as electrons.
- Majority of the atom is empty space. If nucleus were the size of a pencil eraser, the closest electron would be 100 yards away!

Chapter 16

Video by Madelyn Lloyd, Julianna Sorvillo, Destiny McClain, and Taren Chism on Chapter 16 Section 1.

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Objectives 16.1.1 Explain
how heat and work transfer
energy. 16.1.2 Relate

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Answers to the motion of particles that make up a material. 16.1.3 Relate temperature to thermal energy and to thermal expansion. 16.1.4 Calculate thermal energy, temperature change, or mass using the specific heat equation. 16.1.5 Describe how a calorimeter

Section 16.1 16.1 Thermal Energy and Matter

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