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12. Inert gasses are arranged in what part of the Periodic Table? Far right group or column. PART III - Chemical Reactions, Reactants and Products. 13. A liquid that forms a gas when heated is an example of a chemical reaction. False 14. When Vinegar (acetic acid  $\text{CH}_3\text{COOH}$ ) reacts with baking soda ( $\text{NaHCO}_3$ ), one of the products is Carbon Dioxide ( $\text{CO}_2$ ). True 15.

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molecule on the reactant side. Because we added HCl and NaOH in steps 1-4, the colours changed back to their original shades to maintain equilibrium. 7) BaCr<sub>2</sub>O<sub>7</sub> is much more soluble than BaCrO<sub>4</sub>, as the K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> solution was darker in colour when mixed with Ba(NO<sub>3</sub>)<sub>2</sub>. 8) No precipitate formed in step 7 of Part IV because H in HCl was added, shifting the  $\rightleftharpoons$  reaction to the reactant side.

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Keith began writing for the Fool in 2012 and focuses primarily on healthcare investing topics. His background includes serving in management and consulting for the healthcare technology, health ...

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