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There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can refer you to somebody.) Those three states are solid, liquid, and gas. These three states are

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quite different. The main difference is in their particles.

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Section 13.4

Changes of State

The relationship among the solid, liquid, and vapor states (or phases) of a substance in a

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Sealed container
are best
represented in a
single graph called
a phase diagram
Phase diagram-
gives the
temperature and
pressure at which a
substances exists
as solid, liquid, or
gas (vapor) 70.

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of Matter" for Mr.
Testa's Honors
Chemistry class.
Still a work in
progress. Currently
only contains some
gas laws and
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SECTION 13.1 THE
NATURE OF GASES

(pages 385–389)

This section

introduces the

kinetic theory and

describes how it

applies to gases. It

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defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance. Kinetic Theory and a Model for Gases (pages 385–386) 1.

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