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Chapter 13 Properties of Solutions Chemistry, The Central Science , 10th edition Theodore L. Brown; H. Eugene LeMay, Jr.; and Bruce E. Bursten. Solutions Solutions • Solutions are homogeneous mixtures of two or more pure substances. • In a solution, the solute is dispersed uniformly

[Chapter 13 Properties of Solutions](#)

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Chemistry: Chapter 13: Solutions. solution. suspension. solvent. solute. a homogeneous mixture of two or more substances uniformly disp.... a mixture in which particles of a material are more or less ev.... in a solution, the substance in which the solute is dissolved.

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Chemistry Chapter 13: Solutions. STUDY. PLAY. How can a sugar cube be dissolved quicker in a 1000 ml of water solvent? 1. Break up the sugar cube, to increase the surface area of the solute. This allows more solute to be in contact with the solute, speeding the dissolving process

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This video explains the concepts from your packet on Chapter 13 (Properties of Solutions), which can be found here: <https://goo.gl/etUYcM>
Section 13.1: The S...

Chapter 13 Properties of Solutions - YouTube

Chapter 13 Properties of Solutions. Learning goals and key skills: Describe how enthalpy and entropy changes affect solution formation. Describe the relationship between intermolecular forces and solubility, "like dissolves like". Describe the role of equilibrium in the solution process and relationship to the solubility of a solute. Describe the effect of temperature on solubility of ...

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