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Compounds With Transition Metals

How to Name Chemicals Made Easy

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*to memorize in 30 min* ~~Nomenclature:  
Functional groups Naming Common  
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Isotopes, Percent Abundance, Atomic  
Mass | How to Pass Chemistry Writing  
Chemical Formulas For Covalent  
Molecular Compounds Naming Ionic  
Compounds 7.1 Chemical Names and  
Formulas (2/2) Writing Ionic Formulas  
with Transition Metals Naming Binary  
Ionic Compounds With Transition  
Metals \u0026 Polyatomic Ions  
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## **Chemical Names And Formulas Guide**

The common names and formulas of chemical compounds is a very relevant area of study for every competitive exam including SSC and it is important for the aspirants to develop a fair idea on this subject. In this article, we are discussing the following topics.

### **List of Chemical Compounds and their Common Names and Formulas**

Chemical formula Synonyms CAS

number CaAl 2 O 4: calcium

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aluminate: 12042-68-1 CaB 6: calcium  
boride: 12007-99-7 CaBr 2: calcium  
bromide: 7789-41-5 Ca(BrO) 2:  
calcium hypobromite: Ca(BrO 2) 2:  
calcium bromite: Ca(BrO 3) 2: calcium  
bromate: 10102-75-7 Ca(BrO 4) 2:  
calcium perbromate: Ca(CN) 2:  
calcium cyanide: 592-01-8 CaCO 3:  
calcium carbonate spent lime calcite  
limestone marble: 471-34-1 CaC 2

## **Glossary of chemical formulae - Wikipedia**

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Chemical formulae provide insight into the elements that constitute the molecules of a compound and also the ratio in which the atoms of these elements combine to form such molecules. For example, the chemical formula of water, which is  $H_2O$ , suggests that two hydrogen atoms combine with one oxygen atom to form one molecule of water.

## **Chemical Formula of Common Compounds, Chemical reaction ...**

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## **Chemical Names And Formulas Guide**

Here is a list of ten common bases with chemical structures, chemical formulas, and alternate names. Note that strong and weak means the amount the base will dissociate in water into component ions. Strong bases will completely dissociate in water into their component ions. Weak bases only partially dissociate in water.

## **Names of 10 Bases With Chemical Structures and Formulas**

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